







Here on the surface, oil rests above the water because it is less dense. Will the same thing happen in space?

How does two immiscible fluids interact with each other in space, specifically, oil and water?

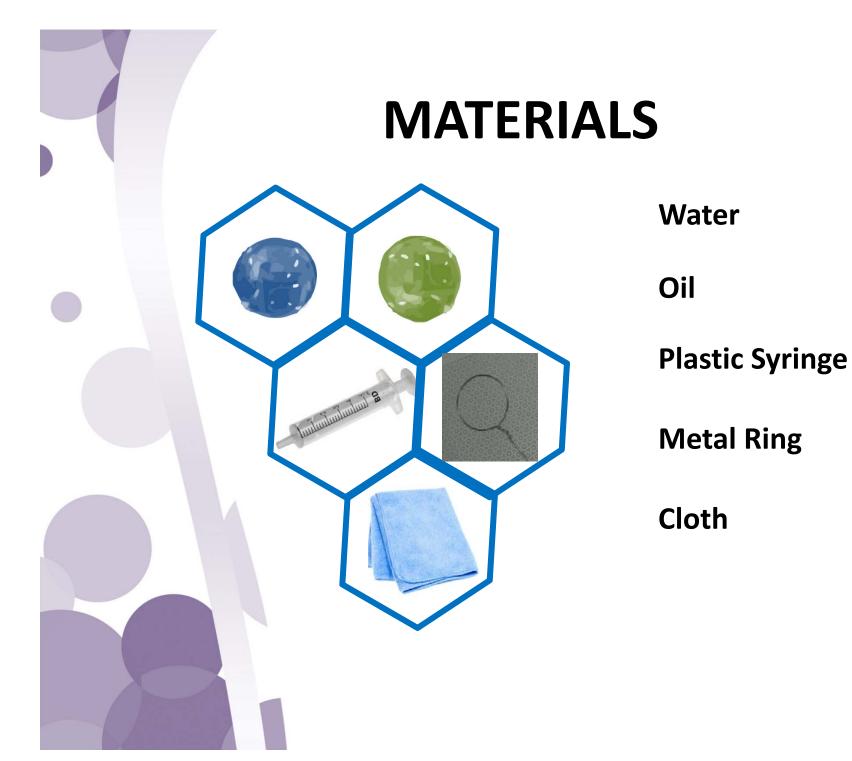


### **HYPOTHESIS**

When oil is injected inside a ball of water in space, it will enclose the water.

If water is injected in oil instead, there would be no change.

On the surface, oil rests on top of the water because it is less dense. In theory, the water should be enclosed by the oil.





## **PROCEDURE**

- 1. Form a ball of water supported inside the metal ring
- 2. Inject oil inside the ball of water using the syringe and observe what happens.





## **PROCEDURE**

- 3. Clean the setup using syringe and cloth
- 4. Redo experiment by forming the ball of oil first and observe what happens
- 5. Clean the setup using syringe and cloth



#### **RESULTS**



In this image, after injecting water in the oil, water (blue) remains inside the ball of silicon oil (colorless).

Images taken from "Double-Layered Liquid Ball" experiment on Asian Try Zero-G 2018 conducted by JAXA astronaut Norishige Kanai at the ISS Kibo Module

#### **RESULTS**



In this image, upon injecting oil in the water, although unclear on initial observations, the silicon oil (colorless) stays inside the ball of water (blue).

Images taken from "Double-Layered Liquid Ball" experiment on Asian Try Zero-G 2018 conducted by JAXA astronaut Norishige Kanai at ISS Kibo Module



Under microgravity conditions, other forces tend to dominate in space. In this experiment, density didn't play a role in the results. Instead, it was surface tension.

Both cases in this experiment show that the liquids have not changed their positions. The water cannot penetrate the oil in the interface and the same is true the other way around.

## References

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# **THANK YOU!**